

EXAMPLE 16-9

Calculating Ion Concentrations in a Polyprotic Acid

Solution. For a 3.0 M H_3PO_4 solution, calculate:

(a) $[\text{H}_3\text{O}^+]$; (b) $[\text{H}_2\text{PO}_4^-]$; (c) $[\text{HPO}_4^{2-}]$ (d) $[\text{PO}_4^{3-}]$

	H_3PO_4	$+$	H_2O	$\rightleftharpoons$	H_2PO_4^-	$+$	H_3O^+
Initial conc.	3.0 M				0		0
Changes	$-x$ M				$+x$ M		$+x$ M
Equilibrium Concentration	$(3.0-x)$ M				x M		x M

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web 2022 február 28 13 51 gyulai tankerületi központ által meghirdetett intézményvezetői pályázatok az emberi erőforrások minisztere a közalkalmazottak jogállásáról szóló 1992 évi xxxiii törvény 20 a a alapján az alábbi intézményvezetői magasabb vezetői pályázatokat hirdeti meg bucsai ii