# PROBLEMS, DEVELOPMENT TRENDS, AND CHALLENGES IN PHYSICAL CHEMISTRY

# Interfacial Synthesis: Morphology, Structure, and Properties of Interfacial Formations in Liquid—Liquid Systems

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Abstract—The results of studies in the field of interfacial synthesis and interfacial formations in liquid—liquid systems are summarized. The mechanisms of the processes of interfacial synthesis are considered. Data on the self-assembly of nanoparticles, films, and 3D materials are given. The properties of materials of interfacial formations in systems with rare-earth elements and dis(2-ethylhexyllphosphoric acid, obtained both in the presence and absence of local vibrations, are described. It was established that materials obtained in the presence of local vibrations in the interfacial layer have higher density, melting point, and magnetic susceptibility and lower electric conductivity. The effect of force field parameters on the properties of interfacial formations is considered. Practical applications and prospects for research in the field of interfacial formations are discussed.

Keywords: vibrational action, interfacial synthesis, nanomaterials, self-assembly, rare-earth element DOI: 10.1134/S0036024421040075

Interfacial synthesis proceeds in heterophase liquid systems by a chemical reaction between substances that are initially in different liquid phases. The reaction can take place on a flat surface that separates immiscible liquids, or on a nonflat surface in microemulsions. There is a detailed review on microemulsions [1]; therefore, reactions in microemulsions are not discussed in the present work. An interfacial reaction can be polymerization, which has long been used to obtain polymer films. The present review considers only those publications on polymerizations that reported the preparation of nanomaterials [2, 3].

As a result of a heterophase reaction, the molecules of a new substance are formed at the liquid-liquid interface and can be localized at the interface or distributed in the liquid. The molecules localized at the interface can form nanoparticles (NPs); in this case, nanomaterials form by the "bottom-up" method. The liquid—liquid interface promotes self-organization and self-assembly of nanoparticles (NPs). The selfassembly of NPs was widely discussed in the literature. Reviews [4-10] described the approaches to the creation of ordered structures of nanoparticles, gave examples of formation of these structures, and analyzed the driving forces of self-organization and the physicochemical properties of ordered assemblies. The authors of [4] underlined the importance of understanding the driving forces of interfacial assembly, seeking new methods for use as probes for interfacial assembly, and controlling particle interactions and the possibility of external effects. The fundamentals and applications of liquid—liquid interface for the creation of complex fully liquid devices with many potential applications were reviewed in [7]. Review [8] focused on modification of the interface with assemblies of gold nanoparticles or nanofilms. The behavior of self-assemblies of biological and synthetic particles was discussed in [9]. According to the authors, the interfacial assembly of single-walled carbon nanotubes at liquid interfaces will play a key role in some applications such as fractionation of nanotubes, fabrication of thin films, and synthesis of porous foam plastics and polymer composites. The use of functionalized interfaces of two immiscible electrolyte solutions in electrocatalysis and electroanalysis of modified boundaries was considered in [10].

Interest in interfacial formations in systems with organic acids and salts of a and f elements was dictated by their practical utility and prospects for production of nanomaterials with desired properties on their basis. Moreover, the properties of these materials depend on the conditions of their preparation, in particular, on the presence of an external force field that can affect both the interfacial synthesis and the self-assembly of nanoparticles at the interface.

The effects of the vibration field on the interface synthesis and interface formations were considered in [11, 12], but not reviewed. Here we made an attempt to fill this gap. The review presents the interfacial syntheses of substances that give metal nanoparticles, metal

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