$M\Box T\Box W\Box T\Box F\Box S\Box$ Date:...../20...... [Cu(H20)4] CU (NH3)4 Rate = Kobs (CU (H20)) NH3)4 steps not in a single step. This reaction complete in (CU (H20)4) + MH3 C4 (H20) LNH3 We said that It proceeds in K3 LNHS usless due to four CU(H20), (NH3) ligands replacement. CU (M20) (NH3)3 we consider a spaction between co(iii) cr(Ti) in the presence of Ag(I) as a autalyst and perchloric acid is used as solvent. ADCE) we said the coil + Cx(tin) -(H) - (() perchloric (VI) awin e Pate = kabs CO(10)] (CY (0)] [Ag (5)] THIOSES. empirical formula. electron and la mstep

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