

# Naming Inorganic Compounds

## metal names

Lower charge gets the ous

Higher charge gets the ic

$\text{Fe}^{2+} = \text{Fe(II)} = \text{Ferrous}$

$\text{Fe}^{3+} = \text{Fe(III)} = \text{Ferric}$

$\text{FeCl}_2 =$  ferrous chloride

$\text{FeCl}_3 =$  ferric chloride

$\text{Cu}^+ = \text{Cu(I)} = \text{cuprous}$

$\text{Cu}^{2+} = \text{Cu(II)} = \text{cupric}$

$\text{CuBr} =$  cuprous bromide

$\text{CuBr}_2 =$  cupric bromide

$\text{Hg}^+ = \text{Hg(I)} = \text{Mercurous}$

$\text{Hg}^{2+} = \text{Hg(II)} = \text{Mercuric}$

$\text{HgI} =$  mercurous iodide

$\text{HgI}_2 =$  mercuric iodide

# How To Name An Inorganic Substance

**Sam Stuart**



## **How To Name An Inorganic Substance:**

**How to Name an Inorganic Substance** Sam Stuart, 2013-10-22 How to Name an Inorganic Substance serves as a guide to the use of nomenclature of inorganic chemistry. This book contains a few references to the rules for the nomenclature of organic chemistry as well as of inorganic boron compounds. This text defines inorganic compounds as substances consisting of combinations of all the elements except those that comprise mainly of certain chains and rings of carbon atoms with defined atoms and groups attached to these skeletal atoms. This book presents as well the background principles involved in or related to nomenclature including oxidation number, coordination number, multiplying affixes, enclosing marks, and use of italic letters. This guide also explains the various types of names used in chemistry including substitutive names, simple binary names, molecular hydride or anion names, and cations derived by proton addition to molecular hydrides. This book is a valuable resource for organic and inorganic chemists. [How to Name an Inorganic Substance](#) International Union of Pure and Applied Chemistry. Commission on the Nomenclature of Inorganic Chemistry, 1977

**How to Name an Inorganic Substance: Being a Guide to the Use of Nomenclature of Inorganic Chemistry: Definitive Rules 1970**, 1977

**How to Name an Inorganic Substance**, 1977 **Nomenclature of Inorganic Chemistry** Sam Stuart, 2013-09-11 Nomenclature of Inorganic Chemistry Second Edition deals with the nomenclature of boron hydrides and higher hydrides of the Group IV-VI elements, organometallic compounds, and polyacids. This book deals with organoboron, organosilicon, and organophosphorus compounds. Organized into 11 chapters, this edition begins with an overview of the concept of oxidation number and coordination number as well as the conventions governing the use of multiplying affixes, enclosing marks, letters, and numbers. This text then discusses the standardization of the formula of inorganic compounds to demonstrate the structural connections between atoms and to provide other comparative chemical information. Other chapters consider nomenclature for radicals and ions. This book discusses as well the nomenclature for binary and pseudobinary acids, oxoacids, peroxyacids, and chloroacids. The final chapter deals with the nomenclature for boron hydrides, boron radicals, and anions and cations derived from the boranes. This book is a valuable resource for organic and inorganic chemists. *How to Name an Inorganic Substance*, 1977

**Inorganic Substances** Derek William Smith, 1990-01-18 Inorganic Substances is complementary in its approach to conventional inorganic chemistry textbooks. Written with the undergraduate in mind, it gives an introduction to descriptive inorganic chemistry, a systematic survey of the chemistry of the elements according to the Periodic Classification. In this way, the reader acquires a firm grasp of the principles which underlie which inorganic substances can be made, their preparations, structures, chemical reactions, and physical properties. The book presents theory as a background to the facts of inorganic chemistry rather than as an end in itself. It does not concentrate on structural detail or reaction mechanisms but stresses the interplay between thermodynamic and kinetic considerations in understanding stability. The ways in which the various theories of structure and bonding are related are thoroughly dealt with throughout.

The approach of this book makes it a useful companion to any intermediate inorganic chemistry course. It should also be useful to other science students especially earth and material scientists who need a good grounding in modern inorganic chemistry.

*Nomenclature of Inorganic Chemistry* International Union of Pure and Applied Chemistry, 2005. The Red Book is the definitive guide for scientists requiring internationally approved inorganic nomenclature in a legal or regulatory environment.

**How to Name an Inorganic Substance** International Union of Pure and Applied Chemistry. Commission on the Nomenclature of Inorganic Chemistry, 1977.

*How to Name an Inorganic Substance* International Union of Pure and Applied Chemistry, 1977.

**Rules for Naming Inorganic Compounds** B. Smith Hopkins, 1941.

**Handbook of Terminology Management**, 1997-03-21. The Handbook of Terminology Management is a unique work designed to meet the practical needs of terminologists, translators, lexicographers, subject specialists, e.g. engineers, medical professionals, etc., standardizers and others who have to solve terminological problems in their daily work. In more than 900 pages, the Handbook brings together contributions from approximately 50 expert authorities in the field. The Handbook covers a broad range of topics integrated from an international perspective and treats such fundamental issues as practical methods of terminology management, creation and use of terminological tools, terminology databases, on-line dictionaries, etc., terminological applications. The high level of expertise provided by the contributors combined with the wide range of perspectives they represent results in a thorough coverage of all facets of a burgeoning field. The layout of the Handbook is specially designed for quick and for cross reference with hypertext and an extensive index. See also Handbook of Terminology Management, set volumes 1 and 2.

[How to Name an Inorganic Substance](#) International Union of Pure and Applied Chemistry, 1975.

*Chemical Nomenclature* K.J. Thurlow, 2012-12-06. Chemical nomenclature can be a complicated subject. As a result, most works on the subject are rather dry textbooks and primarily consist of sets of instructions on how to name chemicals. This practical book proves that chemical nomenclature can be interesting, not just a necessary evil. Written in a lively and engaging style by experts in their particular fields, this new book provides a general discussion on why good, clear nomenclature is needed. It introduces the reader to the various forms of nomenclature without reading like a textbook. Both systematic and trivial nomenclature systems are used widely and interchangeably in chemistry, and this new book covers both areas. For example, systematic nomenclature in both the CAS and IUPAC styles is introduced. These systems have many similarities but important differences which the chemist should be aware of. Specialized naming systems are needed for polymers and natural products, and these areas are covered in separate chapters. The naming of elements is a very topical subject at the moment, and so this is included to ensure a comprehensive coverage. Covering a wide range of topics in the area of nomenclature and acting as an introduction to a varied field, this book will be of interest to industrial chemists as well as students at senior undergraduate and postgraduate level.

[Systematic Nomenclature of Organic, Organometallic and Coordination Chemistry](#) Ursula Bünzli-Trepp, 2007-01-01. For the first time, chemists, biochemists, pharmacologists, scientists at

all levels in both academia and industry documentalists editors and software developers can rely on a user friendly book which contains everything required for the construction or interpretation of systematic names of organic organometallic or coordination compounds as well as those for more complicated molecules      Chemical Principles Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding      **CRC Handbook of Chemistry and Physics** William M. Haynes,2016-04-19 Mirroring the growth and direction of science for a century the Handbook now in its 93rd edition continues to be the most accessed and respected scientific reference in the world An authoritative resource consisting tables of data its usefulness spans every discipline This edition includes 17 new tables in the Analytical Chemistry section a major update of the CODATA Recommended Values of the Fundamental Physical Constants and updates to many other tables The book puts physical formulas and mathematical tables used in labs every day within easy reach The 93rd edition is the first edition to be available as an eBook      **How to Name an Inorganic Substance. Being a Guide to the Use of** International Union of Pure and Applied Chemistry (Oxford),1977      Chemical Abstracts ,1921      **Foundations of College Chemistry, Alternate** Morris Hein,Susan Arena,2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

## **How To Name An Inorganic Substance** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**How To Name An Inorganic Substance**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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## **How To Name An Inorganic Substance Introduction**

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