



Chart 43.—Exponential Gas Changes. Relation between Initial and Final Ratios of Pressures, Volumes, Temperatures, and Entropies.

Handbook Of Thermodynamic Tables Diagrams

K. Raznjevic



Handbook Of Thermodynamic Tables Diagrams:

Handbook of Thermodynamic Tables and Diagrams Charles Edward Lucke, 1915 **Handbook of Thermodynamic Tables and Diagrams** Charles Edward Lucke, 2017-12-24 Excerpt from Handbook of Thermodynamic Tables and Diagrams A Selection of Tables and Diagrams From Engineering Thermodynamics The Mollier total heat entropy diagram for steam makes possible the solution of many problems involving both saturated and superheated steam Since this chart is so convenient for turbine work a scale of corresponding steam jet velocities has been added to the diagram Temperature entropy and Mollier diagrams have also been plotted for ammonia and carbon dioxide from which the work may readily be obtained About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works *HANDBK OF THERMODYNAMIC TABLES* Charles Edward 1876-1951 Lucke, John Joseph 1862-1926 Flather, 2016-08-26 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work was reproduced from the original artifact and remains as true to the original work as possible Therefore you will see the original copyright references library stamps as most of these works have been housed in our most important libraries around the world and other notations in the work This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work As a reproduction of a historical artifact this work may contain missing or blurred pages poor pictures errant marks etc Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant **Handbook of Thermodynamic Tables and Diagrams; a Selection of Tables and Diagrams From Engineering Thermodynamics** Charles Edward Lucke, John Joseph Father, 2022-10-27 **Handbook of Thermodynamic Tables and Diagrams** Charles Edward Lucke, John J. Flather, 2014-06-01 An excerpt from the beginning of PART I INTRODUCTION THE province of Engineering Thermodynamics is to guide numerical thermal computations dealing with actual substances and apparatus in accordance with the laws of thermodynamic philosophy In order to do this numerical values for heat effects must be available for the various substances and materials used in engineering under the varying conditions of practice and in such units as may readily be applied these include especially that class of units known as physical constants which embrace for example such quantities as the coefficients of expansion the specific heats latent heats

of fusion and vaporization the ratio of the pressure volume product to absolute temperature the exponent s in adiabatic expansion of gases and vapors and various other quantities In addition to the physical constants which are necessary in the work of thermodynamic computation the solution of numerical problems is greatly facilitated by the use of other correlated tables and diagrams many of which are given in the present book of tables but to correctly use such aids there should be no ambiguity in regard to the units employed It should be noted that true pressures are always absolute that is measured above a perfect vacuum or counted from zero while most pressure gages and other devices for measuring pressure such as indicators give results measured above or below atmospheric pressure In all problems involving work of gases and vapors the absolute values of the pressures must be used hence if a gage or indicator measurement is being considered the pressure of the atmosphere found by means of the barometer must be added to the pressure above atmosphere in order to obtain the absolute or true pressures When the pressures are below atmosphere the combination with the barometric reading will depend on the record if the record be taken by an indicator it will be in pounds per square inch below atmosphere and must be subtracted from the barometric equivalent in the same units to give the absolute pressure in pounds per square inch When however a vacuum gage reads in inches of mercury below atmosphere as such gages do the difference between its reading and the barometric gives the absolute pressure in inches of mercury directly which can be converted to the desired units by the proper factors In general steam pressures are most commonly stated in pounds per square inch and are designated as either gage or absolute Pressures of compressed air are commonly expressed in the same units as steam either gage or absolute though sometimes in atmospheres Steam pressures below atmosphere are conveniently stated as a vacuum of so many inches of mercury or they may be given as a pressure of so many inches of mercury absolute or so many pounds per square inch absolute The pressures of gases stored in tanks under high pressure are frequently recorded in atmospheres due to the convenience of computation of quantities on this basis Pressures of air obtained by blowers or fans are sometimes given in ounces per square inch above atmosphere but such pressures and also differences of pressure of air due to chimney draught or forced draught and the pressure of illuminating gas in city mains are commonly stated in inches of water In many cases the data are given in other units which must be converted by the use of tables diagrams or otherwise before the results can be properly interpreted or intelligently compared Time is an important item in all engineering work and none the less so in computations so that convenient tables and diagrams are most essential to the solution of such problems

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