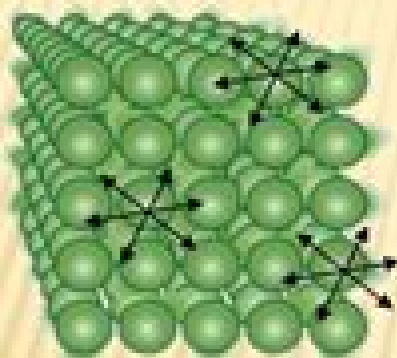


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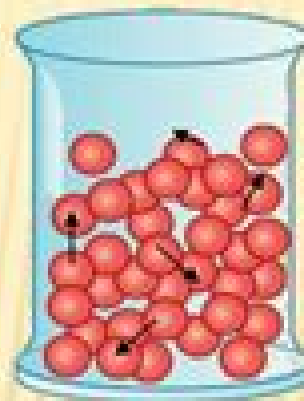
Kinetic Theory explains the differences between the three states of matter.

solid



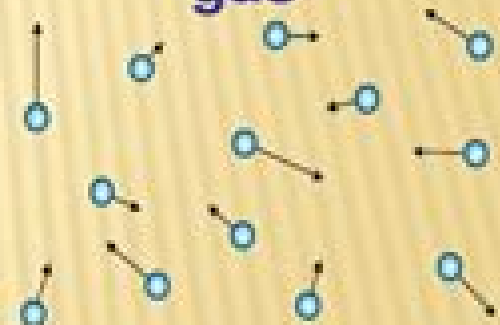
In solids, the particles have low kinetic energy and they are held together by intermolecular forces of attraction.

liquid



In liquids, the particles have more kinetic energy to stretch the intermolecular forces of attraction.

gas



In gases, the particles have enough kinetic energy to break all intermolecular forces of attraction.

Kinetic Theory Of Liquids

Richard D Kleeman



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