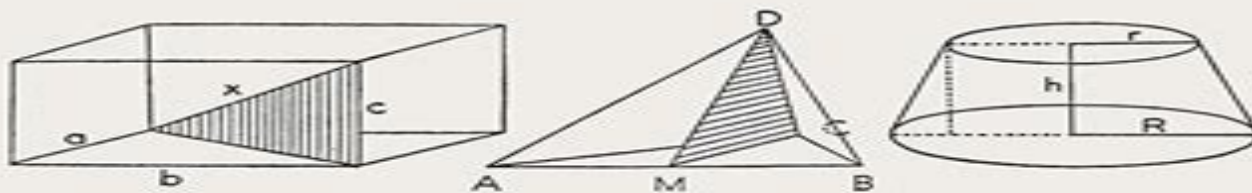


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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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web mar 6 2017 1 a beryllium atom has 4 protons 5 neutrons and 4 electrons what is the mass number of this atom 4 5 8 9
13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5
neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the
molecular structure are trigonal bipyramid c both the electron geometry and the molecular structure are linear d both the
electron geometry and the molecular structure are trigonal planar

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web a an oxygen atom with 8 protons and 8 neutrons b a potassium atom with 19 protons and 20 neutrons c a lithium atom
with 3 protons and 4 neutron give complete symbols of each atom including the atomic number and the mass number

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proton and neutron in the structure of an atom the nucleus sets in the center and it contains the proton whereas electrons lie
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all matter the mass relationships between elements and compounds in chemical reactions ultimately relate back to the
characteristics of the atoms of which they are composed

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of an element what are the three main subatomic particles in an atom the three main subatomic particles in an atom are
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