

## Chapter 1: Learning objective / Guide

The learning objectives provided below are designed to help you prepare for exam. These goals are in no way intended to be all inclusive. You are responsible for all materials covered in the course.

### Chapter 1: Matter, Energy, and Measurement.

1. Define and use terms such as matter, chemical change, and physical change

Chemistry is the study of matter

- **Matter** is anything that has mass and takes up space.  
Matter can change from one form to another.
- In a **chemical change** (chemical reaction), substances are used up and others formed in their place.  
Example: When propane (bottled or LP gas) burns in air, propane and oxygen are converted to carbon dioxide and water.
- In a **physical change**, matter does not lose its identity.  
A common physical change is a change of state.  
Example: Ice (solid water) melts to become liquid water, liquid water boils to become steam (gaseous water).

2. Be able to Distinguish between a fact, a hypothesis and a theory, *show how these are related to the scientific method and the experimental process.*

**Fact:** A statement based on direct experience

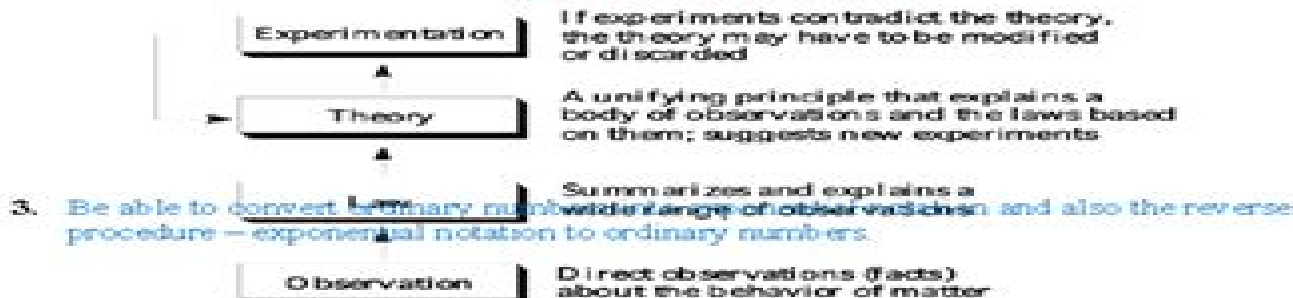
**Hypothesis:** A statement that is proposed, without actual proof, to explain a set of facts or their relationship.

**Theory:** The formulation of an apparent relationship among certain observed phenomena, which has been verified to some extent.

*A theory is the same as a hypothesis except that we have a stronger belief in it because more evidence supports it.*

*If, however, we find new evidence that conflicts with the theory, it must be altered or rejected.*

### SCIENTIFIC METHOD



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**IM Harris**



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