ANSWERS TO THE SHORT QUESTIONS

GENERAL FEATURES OF ALKYL HALIDES

- Q.1 What are alkyl halides? Give their general formula and name two monohaloalkenes. (Sarg. 2010)
- Ans. Those organic compounds which are derived from the hydrocarbons by the replacement of one or more hydrogen atoms by halogen atoms are called alkylhalides. They are represented by R-X and the general formula is C_nH_{2n+1} -X. CH_3Cl , C_2H_5Br , C_3H_7l are mono halo alkanes.
- Q.2 What are primary, secondary and tertiary alkyl halides? Give examples of each. (Faisalabad 2008, Multan 2012, Multan 2013, D.G. Khan 2013)
- Ans. In primary alkyl halides, α-carbon has at least two hydrogen atoms, in secondary alkyl halides, α-carbon has only one hydrogen and in tertiary alkyl halides α-carbon has no hydrogen.

$$CH_3$$

 $CH_3 - CH_2 - Cl (p^\circ)$ $CH_3 - CH - Cl (s^\circ)$ $CH_3 - CH_3 - CH_3$
 CH_3 CH_3 CH_3

PREPARATIONS OF ALKYL HALIDES

- Q.3 What are the necessary conditions to convert alcohols into alkyl halides by using halogen acids? (Lahore 2008, Lahore 2010, Lahore 2012, F. Abad 2013)
- Ans. Alcohols are reacted with halogen acids in the presence of ZnCl₂ or H₂SO₄ to get alkyl halides,

$$R - OH + H$$
 $ZnCl_2 \text{ or } H_2SO_4 \longrightarrow R - X + H_2O$

- Q.4 SOCl₂ is the best reagent to get alkyl halides from alcohols. Which solvent is necessary to complete this reaction?

 (Lahore 2008, Guj. 2010, Lahore 2012)
- Ans. Alcohols react with SOCl₂ to give alkyl cholrides in the presence of pyridine. HCl and SO₂ are gaseous products which are evolved automatically and pure alkyl chloride is obtained. So, this is one of the best methods.
- Q.5 Convert to get alkyl chlorides to get alkyl chlorides ethyl alcohol into their respective halides by using PCl₃ and PI₅. (Lahore 2008, Lahore 2012)

Ans.
$$3CH_3CH_2OH + PCI_3 \longrightarrow 3CH_3CH_2CI + H_3PO_3$$

 $CH_3CH_2OH + PI_3 \longrightarrow CH_3CH_2I + POI_3 + HI$
REACTIONS OF ALKYL HALIDES

- Q.6 Which two factors are responsible for reactivity of alkyl halides as compared to alknes? (Faisalabad 2010, Guj. 2012)
- ns. (i) Bond dissociation energy (ii) Polarity

The bond dissociation energy of C - X bond is in the order.

$$C - F > C - Cl > C - Br > C - I$$

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