

ocyanic acids - Fractional Most

- Carlo Exerc. Acids.

#### Chapter 16: Acids and Bases Learning goals

- Define acids and bases and compare and contrast th Lowry acid-base models
- · Identify and predict conjugate acid-base pairs
- Calculate [H3O+] and [OH-] based on the autoionization of water
- Identify solutions as acidic/basic/neutral
- · Relate acid and base strength to Ka values
- Calculate [H3O+], [OH-], pH, and pOH of a strong acid or base solution, and convert pH and pOH
- Calculate the pH of a weak acid or base solution using an ICE table and the acid dissociation constant

Note: We are not including percent ionization or mixtures of acids (subsections of 16.7)

## Characteristics of acids and bases (16.2)

BRICH STRUCTURE

Access acids

#### Acids

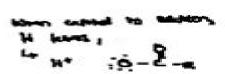
- sour taste
- dissolve metals
- neutralize bases
- · turn blue litmus paper red

#### Bases

- Bitter taste.
- Slippery
- Neutralize acids

q. Carboxylic acids:

Turn red litmus paper blue



# <u>Lecture Outlines For General Chemistry Principles And Structure</u>

**Columbia University. Teachers College** 

## **Lecture Outlines For General Chemistry Principles And Structure:**

Lecture Outlines for General Chemistry Ronald O. Ragsdale, James E. Brady, Gerard E. Humiston, 1986-04-01 General Chemistry James E. Brady, Gerard E. Humiston, 1986 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems **Lecture Outline-General Chemistry** Kenneth W. Whitten, Raymond E. Davis, Larry Peck, George G. Stanley, 2006-03 By Charles H Atwood and Kenneth W Whitten both of the University Georgia Richard M Hedges of Texas A M University and revised by Kimberly Schurmeier of the University of Georgia Detailed lecture outlines of all of the text chapters are available to free students from tedious extensive note taking The outlines enable the students to listen more efficiently because they know that the important ideas Lecture Outline to Accompany General Chemistry Charles H. and terms are already written down for them Atwood, Kenneth W. Whitten, 2000 Official Documents, Comprising the Department and Other Reports Made to the Governor, Senate and House of Representatives of Pennsylvania, 1893 **University of Chattanooga Bulletin** University The Temple University Catalogue Temple University, 1913 Calendar University of of Chattanooga, 1911 Saskatchewan, 1918 Catalogue Kansas State Agricultural College, Kansas State College of Agriculture and Applied **Program** Bryn Mawr College, 1905 Science, Kansas State University, 1922 Drexel Institute Yearbook ,1900 Bulletin... Detroit (Mich.) College of the city of Detroit,1927 Scientific and Technical Books and Serials in Print ,1989 Catalog ... St. John's University (Collegeville, Minn.),1909 **Notes and Queries: A Medium of** Inter-Communication for Literary Men, Artists, Antiquaries, Genealogists, Etc., 1859 **Annual Report of the** Pennsylvania Department of Agriculture Pennsylvania. Department of Agriculture, 1892 Report of Pennsylvania Forestry Commission published in 1896 1895 pt 2 **Proceedings of the ... Meeting of Veterinary Officers in India ...** India. Board of Agriculture, 1919 Announcement Columbia University. Teachers College, 1928 The Dental Cosmos J. D. White, John Hugh McQuillen, George Jacob Ziegler, James William White, Edward Cameron Kirk, Lovick Pierce Anthony, 1915 Higher structures Mansfield Merriman, Henry Sylvester Jacoby, 1898

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  - Chapter 2: Essential Elements of Lecture Outlines For General Chemistry Principles And Structure
  - Chapter 3: Lecture Outlines For General Chemistry Principles And Structure in Everyday Life
  - Chapter 4: Lecture Outlines For General Chemistry Principles And Structure in Specific Contexts
  - ∘ Chapter 5: Conclusion
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